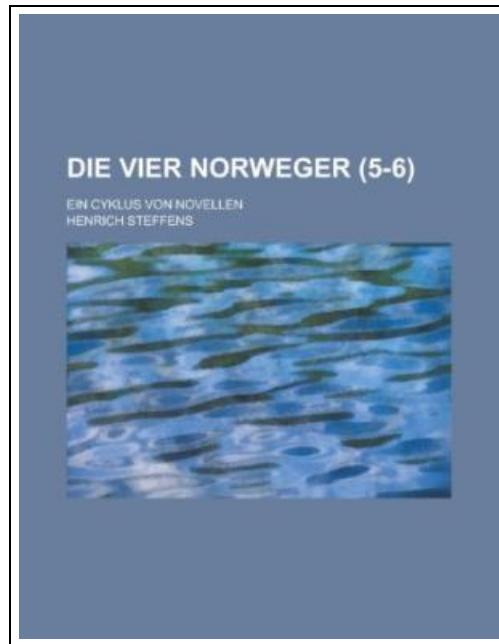


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RareBooksClub. Paperback. Book Condition: New. This item is printed on demand. Paperback. 24 pages. Original publisher: Greenbelt, Md. : NASA Goddard Space Flight Center, 2000 OCLC Number: (OCoLC)83760443 Excerpt: . . . 9. Data Manipulations 9. 1 Formulae Rf Cstd Astd Cs Rf As Where: Rf Response factor Astd Average of 10 standard peak areas Cstd Concentration of the standard Cs Concentration of the sample As Peak area of sample CH4 and CO2 concentrations were calculated from the average of 10 peak areas of known CH4 and CO2 standards. The response factor was calculated as the concentration of the known standard divided by the average of 10 standard peak areas. The peak area of the unknown sample was multiplied by the response factor. Fitting a regression curve to the time series of Cl-I4 and CO2 concentrations made the flux calculations. The flux rate of a gas is calculated using the following equation:  $F = \frac{P R MW}{8.2054 \times 10^5} \frac{T}{V_c} \frac{A_c}{C}$  Where: F flux ( mg m<sup>2</sup> d ) P Chamber pressure ( atm ) R 8.2054 x 10<sup>5</sup> m<sup>3</sup> atm mol K MW Molecular weight of the analyte T Chamber temperature ( K ) V<sub>c</sub> Chamber volume ( m<sup>3</sup> ) A<sub>c</sub> Chamber area ( m<sup>2</sup> ) C ( 1000 mg g<sup>-1</sup> 1440 mirr<sup>d</sup> ) Gases: CH4 16 g mol CO2 44 g mol At sites where oxidation of CH4 occurs, the flux starts at or near ambient CH4 concentrations and is depleted below ambient concentrations. If the correlation coefficients were not significantly high, the flux was determined to be below the detection limit and was regarded as a zero flux rate. If there was an efflux of Cl-h was measured at any of these sites, those data were eliminated. For sites where CH4...

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